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Define Thermodynamic Equilibrium? A system is said to be in thermodynamically equilibrium when there is no change in any macroscopic property of the system. i.e. the temperature and pressure at all points should be same, there should be no velocity gradient there should be no chemical reactions. 3.

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ME8391-Engineering Thermodynamics (ET) is the Anna University Regulation 2017 3rd Semester Mechanical Engineering (Mech) subject. AUNewsBlog team shared some of the useful important questions collection.

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ChE 301 Chemical Engineering Thermodynamics II Che ...

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(ME8391 Important Questions Engineering Thermodynamics) 3. Define flow energy and enthalpy. 4. For a stationary system of fixed mass undergoing a process such that its volume remains constant, Q12 U(T/F) 5. $dQ dh vdp$ for closed system undergoing a process (T/F). (ME8391 Important Questions Engineering Thermodynamics) 6.

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ME6301 ENGINEERING THERMODYNAMICS QUESTION BANK. Two mark questions and answers UNIT II. UNIT II SECOND LAW. 1. Define Clausius statement. ... It is an important thermodynamic property of the substance. It is the measure of molecular disorder. It is denoted by S. The measurement of change in entropy for reversible process is obtained by the ...

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